

06/08/2021.

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Inomière techno.

$$n(C_6H_{12}O) = 6 \times n(C) + 12 \times n(H) + n(O).$$

$$n(H_2O) = 2 \times n(H) + n(O)$$

$$n(H_2O) = m_{\text{entière}}(H_2O) \times N_A \rightarrow 6,02 \times 10^{23}.$$